TECHNICAL NOTES

much more under surface tension that do short, J. Phys. Chem. Solids Vol. 28, pp. 350-351. thick samples; it is possible that this contraction compensated the void formation. An attempt was made to polish some of our samples to look for voids, but the results were inconclusive; voids were certainly not present in such high concentrations on the cathode side as was the case with silver. Some voids were observed in GRONE's samples⁽¹⁰⁾ but were not convincingly distributed. Recent work by HERZIG⁽¹¹⁾ with stacks of copper foils indicated that impure copper shows a reversal whereas pure copper does not.

In conclusion, it appears that there is no intrinsic reversal in marker motion at high temperatures for pure copper. There was no tendency for the marker velocity to decrease in the hot regions, or even to slacken its exponential increase with temperature; the maximum temperatures of this experiment were well above those where previous reversals were observed. The reason for the reversal observed in previous work is still not clear.

Acknowledgements-The author wishes to thank Mr. FELIX DUPONT for assistance in the computations, and is indebted to Dr. H. B. HUNTINGTON for suggesting the work and making the research possible.

Department of Physics GEORGE A. SULLIVAN Rensselaer Polytechnic Institute Troy, New York

References

- 1. WEVER H., Z. Elektrochem. 60, 1170 (1956).
- 2. GRONE A. R., J. Phys. Chem. Solids 20, 88 (1961).
- 3. Kuz'MENKO P. P., Ukr. Fiz. Zh. 7, 117 (1962).
- 4. SULLIVAN G. A., Bull. Am. Phys. Soc. 11, 184
- (1966); Phys. Rev., to be published. 5. SIMMONS R. O. and BALLUFFI R. W., Phys. Rev. 129, 1533 (1963).
- 6. HUNTINGTON H. B. and GRONE A. R., J. Phys. Chem. Solids 20, 76 (1961); HUNTINGTON H. B. and Ho S. C., J. Phys. Soc. Japan 18, suppl. 2, 202 (1963).
- 7. PENNEY R. V., J. Phys. Chem. Solids 25, 335 (1964).
- 8. KUPER A., LETAW H. JR., SLIFKIN L., SONDER E. and TOMIZUKA C. T., Phys. Rev. 96, 1224 (1954). 9. Ho P. S. C., Ph.D. Thesis, Rensselaer Polytechnic
- Institute, 1964; Ho. P. S. and HUNTINGTON H. B, J. Phys. Chem. Solids, 27, 1319 (1966).
- 10. GRONE A. R., Hudson Valley Community College, Troy, N.Y. (Private communication).
- 11. HEUMANN TH., Munster University, Munster, West Germany (Private communication).

Lattice dynamics and the NaCl-CsCl transformation in RbI (Received 20 July 1966)

It has been suggested (1-3) that the spontaneous transformation of RbI from the NaCl to the CsCl structure at high pressure occurs via a structural instability caused by vanishing of the frequency of the transverse acoustic lattice vibrational mode at the center of the $\langle 100 \rangle$ zone face, at a pressure near that of the transformation. This problem is discussed and very much extended by HARDY and KARO.⁽³⁾ Their results indicate that, for appropriate choice of interionic interactions, as the solid is compressed toward the transition conditions the acoustic branches of the lattice vibrational spectrum are strongly shifted to lower energies while the optical branches are shifted slightly to higher energies as shown in Fig. 2 of HARDY and KARO.⁽³⁾ The purpose of this note is to point out that the results quoted are strongly inconsistent with the observed values(4-6) of thermal expansion of RbI if the energy shifts calculated are linear over the pressure range to the transition.

Taking the familiar Quasi harmonic oscillator⁽⁷⁾ point of view the lattice excitations are considered to be those of a set of nearly harmonic oscillators which exhibit a volume dependence of their characteristic energies. This approximation leads to the result for "Grüneisen's Gamma":

$$\gamma_{G} \equiv \frac{\alpha B_{T} V}{C_{v}} = \frac{\sum_{i} \gamma_{i} C_{vi}}{\sum_{i} C_{vi}}$$

in which α is the thermal expansion coefficient, B_T is the Bulk Modulus C_v/V is the heat capacity per unit volume,

$$\gamma_i \equiv \frac{d \ln \nu_i}{d \ln V}$$

a so called "mode gamma", v_i is any lattice vibrational frequency and C_{vi} is the Einstein heat capacity of the i'th mode at the temperature of observation. The summation is over all oscillators making up the crystal vibrational spectrum. At high temperature each oscillator is classically

350

8

03 CO

tin

m

ga

he

us

ga

CU

11

P

20

op

ter

ing

Si

tiv

the

val

coc

ten

tive

the

abe

-

(T

mil

cl

SIC

tic

I. 28, pp. 350-351.

and the NaCI-CsCI tation in RbI NAMES OF T 20 July 1966)

1-3) that the spontaneous rom the NaCl to the CsCl ire occurs via a structural hishing of the frequency of lattice vibrational mode at zone face, at a pressure rmation. This problem is h extended by HARDY and ndicate that, for approprinteractions, as the solid is transition conditions the the lattice vibrational shifted to lower energies nes are shifted slightly to n in Fig. 2 of HARDY and of this note is to point out are strongly inconsistent es⁽⁴⁻⁶⁾ of thermal expanrgy shifts calculated are range to the transition. uasi harmonic oscillator⁽⁷⁾ excitations are considered early harmonic oscillators ne dependence of their This approximation leads isen's Gamma'':

$$=\frac{\sum_{i}\gamma_{i}C_{vi}}{\sum_{i}C_{vi}}$$

expansion coefficient, B_T /V is the heat capacity per

n vi n V

a", ν_i is any lattice vibravi is the Einstein heat de at the temperature of tion is over all oscillators vibrational spectrum. At oscillator is classically

TECHNICAL NOTES

excited and its heat capacity equals the Boltzmann constant k, whence

Y

$$_{\infty} = \frac{\sum_{i} \gamma_{i}}{3N}$$

the simple average of the mode gammas. At intermediate temperatures one sees that the mode gammas are weighted by their associated Einstein heat capacities at the observation temperature. Let us perform a qualitative analysis, estimating mode gammas from Fig. 2 of HARDY and KARO.

Compression data from Slater yields about 4 percent change in volume to the transition pressure. The average shift of the collected acoustic mode peaks is about 15 percent yielding an average acoustic mode gamma -4. Similarly, the average optical mode gamma is about +0.7 yielding a high temperature limiting value of $\Gamma_G = -3.3$ decreasing algebraically with decreasing temperatures. Since in equation (1) B_T , V and C_v must be positive, the implication of the negative gamma is that the thermal expansion coefficient have negative values at all temperatures. The thermal expansion coefficient of RbI has been measured over a wide temperature range.⁽⁴⁻⁶⁾ Its value becomes negative at temperatures below 10°K but even at 5°K the value of γ estimated from the expansion data is about -0.2×10^{-4} , still less negative than value -4 suggested by the lattice dynamical calculations. (The low temperature limit of γ calculated from measurements of the pressure dependence of the elastic constants of RbI is positive.)

BARRON's suggestion⁽⁷⁾ that the thermal expansion of solids is a quantity which yields information about the microscopic interactions in the solid is relevant here. In the present case, it seems that the requirement to account for the experimental values of the thermal expansion and to find a mode frequency which will vanish to permit spontaneous NaCl-CsCl transition will provide a challenge for the lattice dynamicists. We feel at present that the mode instability hypothesis probably is incorrect as the explanation of the spontaneous nature of the transformation on RbI.

This work was supported in part by the Army Research Office (Durham).

Princeton University Princeton New Jersey

W. B. DANIELS

Note added by author in proof-Since submission of this paper, direct measurements have been reported by SAUDERSON, Phys. Rev. Lett. 17, 530 (1966), using inelastic neutron scattering techniques at high pressure, of the shift of the Ta [100] mode energy with compression over the pressure range to the transformation. A decrease of only about 13 percent was observed, which is consistent with the suggestion of the present paper.

REFERENCES

- DANIELS W. B., Bull. Am. Phys. Soc. 7, 237 (1962).
 DANIELS W. B. and SMITH C. S., Physics and Chemistry of High Pressures, Society of Chemical Industry, London (1963).
- 3. HARDY J. R. and KARO A. M., Lattice Dynamics (Ed. R. F. WALLIS) Pergamon Press, Oxford (1965).
- 4. SCHUELE D. E. and SMITH C. S., J. Phys. Chem. Solids 25, 801 (1964).
- 5. WHITE G. K., Proc. R. Soc. A286, 204 (1965).
- 6. SWENSON C. A. Private communication. 7. BARRON T. H. K., Ann. Phys. 1, 77 (1957) and Phil.
- Mag. 45, 720 (1955).